



SUPPLEMENTARY MATERIAL TO

Visible light absorption of TiO₂ nanoparticles surface-modified with vitamin B₆: A comparative experimental and DFT study

TIJANA S. KOVAC¹, ENIS S. DŽUNUZOVIĆ², JASNA V. DŽUNUZOVIĆ³,
BOJANA MILIČEVIĆ⁴, DUŠAN N. SREDOJEVIĆ⁴, EDWARD N. BROTHERS⁵
and JOVAN M. NEDELJKOVIĆ^{4*}

¹Innovation Center, Faculty of Technology and Metallurgy, University of Belgrade,
Karnegijeva 4, Belgrade 11120, Serbia, ²Faculty of Technology and Metallurgy, University of
Belgrade, Karnegijeva 4, 11120 Belgrade, Serbia, ³Institute of Chemistry, Technology and
Metallurgy, Center of Excellence in Environmental Chemistry and Engineering, University of
Belgrade, Njegoševa 12, 11000 Belgrade, Serbia, ⁴Institute of Nuclear Sciences Vinča,
University of Belgrade, P. O. Box 522, 11000 Belgrade, Serbia and ⁵Texas A&M University
at Qatar, P. O. Box 23874, Doha, Qatar

J. Serb. Chem. Soc. 83 (7–8) (2018) 899–909

TABLE S-I. Wavelengths, oscillator strengths, and wave functions of electronic excitations calculated for the [Ti₈O₁₄(OH)₃(B₆)]⁺ clusters

Excited state	Wavelength, nm	Oscillator strength	Wave function (coefficient ² ≥ 10 %)
1	388	0.0002	HOMO → LUMO (70 %)
2	378	0.0003	HOMO-2 → LUMO (26 %) HOMO-1 → LUMO (10 %)
3	369	0.0007	HOMO-6 → LUMO (20 %) HOMO-5 → LUMO (18 %)

*Corresponding author. E-mail: jovned@vin.bg.ac.rs

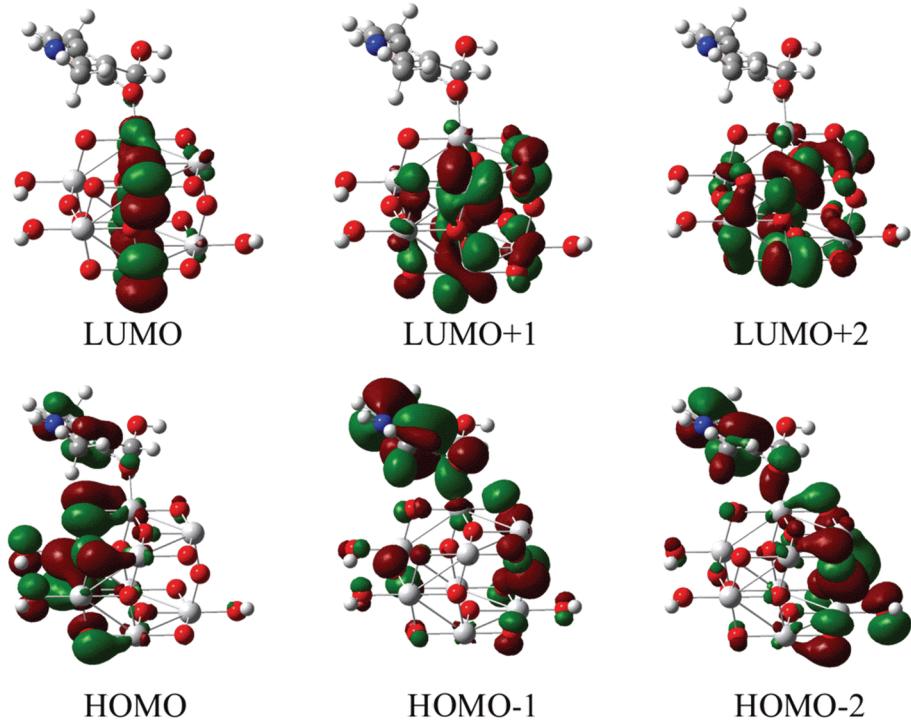


Fig. S-1. Spatial distributions of highest occupied molecular orbitals (HOMO) and lowest unoccupied molecular orbitals (LUMO) for $[\text{Ti}_8\text{O}_{14}(\text{OH})_3(\text{B}_6)]^+$ clusters.