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Supplementary material

#### SUPPLEMENTARY MATERIAL TO

# Continuous flow synthesis of some 6- and 1,6-substituted 3-cyano-4-methyl-2-pyridones

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#### **EXPERIMENTAL**

Preparation of the reactant solutions for synthesis in continuous flow microreactor system

In the first set of experiments, the following solutions were made: acetylacetone (0.06 mol, 6.008 g) and ethyl acetoacetate (0.06 mol, 7.808 g) and added to volumetric flasks, sequentially, then methanol was added up to a volume of 100 mL. The solution of *N*-substituted cyanoacetamide was made in the same way. The corresponding *N*-substituted cyanoacetamide (0.06 mol) was placed in a volumetric flask and deionized water was added up to a volume of 100 mL. Sodium hydroxide pellets (0.07 mol, 2.8 g) were dissolved in deionized water up to a volume of 100 mL.

In the second set of experiments, compounds **1** and **2** were synthesized from the solutions prepared using the following procedure: acetylacetone (0.10 mol, 10.013 g) and ethyl acetoacetate (0.10 mol, 13.014 g) were placed in volumetric flasks, sequentially, and methanol was added up to a volume of 100 mL. Cyanoacetamide (0.15 mol, 12.612 g) and NaOH pellets (0.2 mol, 8 g) were dissolved in deionized water in volumetric flasks up to a volume of 100 mL.

Work-up of the reaction mixture in the continuous flow microreactor system

The reaction mixture assembled in the microreactor was delivered to a test tube containing 1 mL of concentrated HCl. After 9 mL of the mixture was collected, resulting crystals were separated by filtration and washed with deionized water (2 times with 5 mL). Obtained crystals were air dried and analyzed without further purification.

Synthesis under conventional conditions

6- and 1,6-substituted 3-cyano-4-methyl-2-pyridones were prepared from corresponding 1,3-dicarbonyl reagent and N-substituted cyanoacetamides using a modified literature procedure.<sup>1</sup>

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Procedure for the preparation of N-substituted 3-cyano-4,6-dimethyl-2-pyridone in the batch system

Equimolar amounts of acetylacetone and the corresponding N-substituted cyanoacetamide (0.06 mol) were heated under reflux in a water/methanol mixture (120 mL) in the presence of NaOH (0.07 mol) as catalyst for 4 h, except for 3-cyano-4,6-dimethyl-2-pyridone where the reaction time was 1 h. The products were isolated by filtration and purified by crystallization from ethanol.

Procedure for the preparation of N-substituted 3-cyano-6-hydroxy-4-methyl-2-pyridone in the batch system

Equimolar amounts of ethyl acetoacetate and the corresponding *N*-substituted cyanoacetamide (0.06 mol) were heated under reflux in a water/methanol mixture (120 mL) in the presence of NaOH (0.07 mol) as a catalyst for 8 h. The products were isolated by filtration and dissolved in 100 mL of hot water. After cooling, the solution was acidified with concentrated HCl to precipitate the 2-pyridone. The final product was isolated by filtration, washed with deionized water and air-dried.

# CHARACTERIZATION DATA OF THE PRODUCTS OBTAINED IN THE CONTINUOUS FLOW MICROREACTOR SYSTEM

*3-Cyano-4,6-dimethyl-2-pyridone (1).* White powder; m.p.: 285–286 °C (Lit. 290–291 °C<sup>1</sup>); FT-IR (KBr, cm<sup>-1</sup>): 3292 (N–H), 2219 (C–N), 1659 (C=O); <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ , δ / ppm): 2.23 (3H, s, 6-CH<sub>3</sub>), 2.30 (3H, s, 4-CH<sub>3</sub>), 6.17 (1H, s, C5-H), 12.32 (1H, s, OH); UV–Vis (EtOH,  $\lambda_{max}$  / nm): 330.

*3-Cyano-6-hydroxy-4-methyl-2-pyridone* (2). White powder; m.p.: 315–317 °C (Lit. 315–320 °C<sup>2</sup>); FT-IR (KBr, cm<sup>-1</sup>): 3294 (OH), 2223 (CN), 1593 (C=O); <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ ,  $\delta$  / ppm): 2.23 (3H, s, CH<sub>3</sub>), 5.61 (1H, s, C5-H); UV–Vis (EtOH,  $\lambda_{\rm max}$  / nm): 325.

*3-Cyano-1-(2-hydroxyethyl)-4,6-dimethyl-2-pyridone* (*3*). White powder; m.p.: 140–142 °C (Lit. 139–141 °C<sup>3</sup>); FT-IR (KBr, cm<sup>-1</sup>): 2222 (CN), 1663 (C=O), 3268 (OH); <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ , δ / ppm): 2.39 (3H, s, CH<sub>3</sub>), 2.57 (3H, s, CH<sub>3</sub>), 3.71 (2H, m, CH<sub>2</sub>), 4.11 (2H, t, t = 5.4 Hz, CH<sub>2</sub>), 5.04 (1H, t , OH), 6.37 (1H, t , C5-H); UV–Vis (EtOH, t t mm): 334.

3-Cyano-6-hydroxy-1-(2-hydroxyethyl)-4-methyl-2-pyridone (4). White powder; m.p.: 172–174 °C (Lit. 171–172 °C<sup>4</sup>); FT-IR (KBr, cm<sup>-1</sup>): 3367, 3268 (OH), 2223 (CN), 1663 (C=O); <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ , δ / ppm): 2.20 (3H, s, CH<sub>3</sub>), 3.51 (2H, t, J = 6.4 Hz, CH<sub>2</sub>CH<sub>2</sub>OH), 3.99 (2H, t, J = 6,6 Hz, CH<sub>2</sub>CH<sub>2</sub>OH), 5.58 (1H, s, C5-H); UV–Vis (EtOH,  $\lambda_{max}$  / nm): 325.

3-Cyano-4,6-dimethyl-1-propyl-2-pyridone (5). White powder; m.p.: 110–112 °C (Lit. 114 °C¹); FT-IR (KBr, cm⁻¹): 2216 (CN), 1646 (C=O); ¹H-NMR (400 MHz, DMSO- $d_6$ , δ / ppm): 0.98 (3H, t, J = 7.4 Hz, CH<sub>3</sub>CH<sub>2</sub>), 1.67 (2H, m, CH<sub>3</sub>CH<sub>2</sub>), 2.38 (3H, s, 4-CH<sub>3</sub>), 2.53 (3H, s, 6-CH<sub>3</sub>), 3.98 (2H, t, J = 7.8 Hz, CH<sub>2</sub>–N), 6.38 (1H, s, 5-H); UV-Vis (EtOH,  $\lambda_{\text{max}}$  / nm): 324.

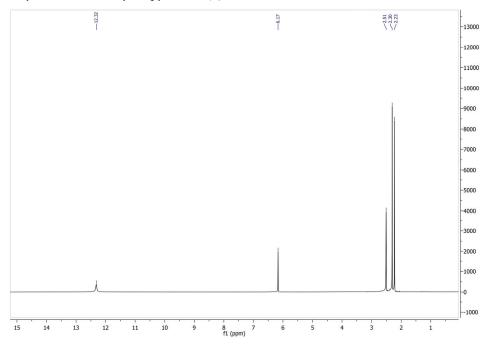
*3-Cyano-6-hydroxy-4-methyl-1-propyl-2-pyridone* (*6*). White powder; m.p.: 238–240 °C (Lit. 239–240 °C<sup>5</sup>); FT-IR (KBr, cm<sup>-1</sup>): 1660 (C=O), 2210 (CN);

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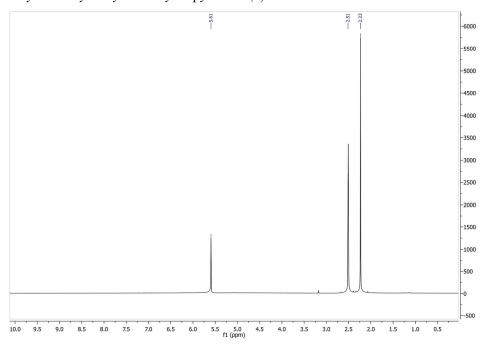
<sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ , δ / ppm): 0.98 (3H, t, J = 7.4 Hz, CH<sub>3</sub>CH<sub>2</sub>), 1.58 (2H, m, CH<sub>3</sub>CH<sub>2</sub>), 2.20 (3H, s, CH<sub>3</sub>), 3.98 (2H, t, J = 7.2 Hz, CH<sub>2</sub>–N), 5.58 (1H, s, 5-H); UV-vis (EtOH,  $\lambda_{\rm max}$  / nm): 325.

#### <sup>1</sup>H-NMR SPECTRA OF THE OBTAINED 2-PYRIDONES

### 3-Cyano-4,6-dimethyl-2-pyridone (1)

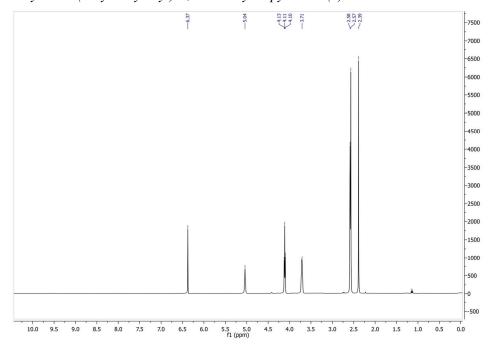


# 3-Cyano-6-hydroxy-4-methyl-2-pyridone (2)

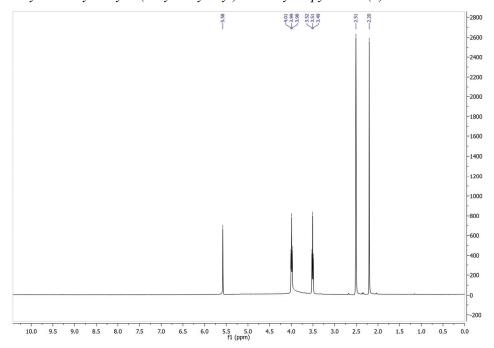


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# $3\hbox{-}Cy ano-1\hbox{-}(2\hbox{-}hydroxyethyl)\hbox{-}4,6\hbox{-}dimethyl\hbox{-}2\hbox{-}pyridone~(3)$

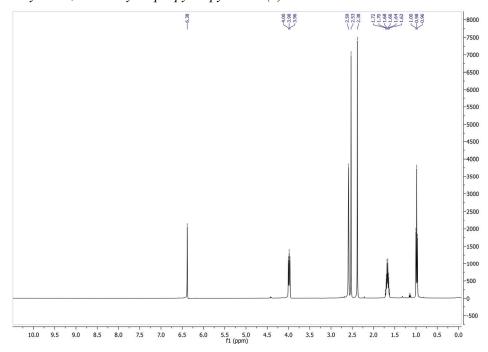


# $3\hbox{-}Cy ano-6\hbox{-}hy droxy-1\hbox{-}(2\hbox{-}hy droxyethyl)\hbox{-}4\hbox{-}methyl\hbox{-}2\hbox{-}pyridone~\textit{\textbf{(4)}}$

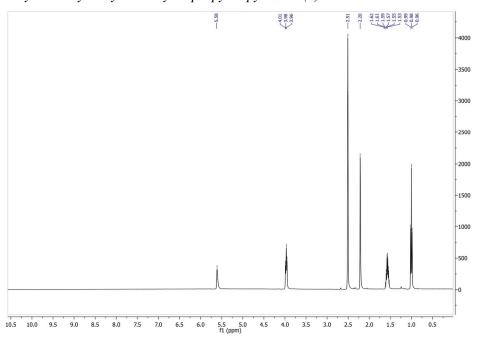


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### *3-Cyano-4,6-dimethyl-1-propyl-2-pyridone (5)*



### 3-Cyano-6-hydroxy-4-methyl-1-propyl-2-pyridone (6)



### REFERENCES

- 1. D. Mijin, A. Marinković, *Synth. Commun.* **36** (2006) 193 (https://dx.doi.org/10.1080/00397910500334421)
- 2. D. Ž. Mijin, M. M. Mišić-Vuković, J. Serb. Chem. Soc. 59 (1994) 959
- 3. A. F. El-Essawy, A. F. Khattab, *J. Heterocycl. Chem.* **41** (2004) 311 (<a href="https://dx.doi.org/10.1002/jhet.5570410302">https://dx.doi.org/10.1002/jhet.5570410302</a>)
- 4. B. D.Tilak, N. R. Ayyangar, U. S. Rao, Indian J. Chem., Sect. B 23 (1984) 18
- 5. S. Balalaie, E. Kowsari, M. S. Hashtroudi, *Monatsh. Chem.* **134** (2003) 453 (https://dx.doi.org/10.1007/s00706-002-0551-2).